# RSC Advances

## PAPER



Cite this: RSC Adv., 2014, 4, 32796

Received 1st May 2014 Accepted 10th July 2014

DOI: 10.1039/c4ra04011b

www.rsc.org/advances

## One-pot route to  $\beta$ -adrenergic blockers via enantioselective organocatalysed epoxidation of terminal alkenes as a key step†

Felix E. Held, Shengwei Wei, Kathrin Eder and Svetlana B. Tsogoeva\*

A convenient and environmentally attractive one-pot two-step process for the synthesis of  $\beta$ -adrenergic blockers via Shi's organocatalytic epoxidation of terminal alkenes and subsequent aminolysis reaction of epoxides with isopropylamine under mild reaction conditions has been developed.

## Introduction

The synthesis of enantiomerically pure chiral drugs and bioactive compounds still remains a challenging task for organic chemists.

Asymmetric syntheses employing chiral catalysts is among the most common routes to optically pure compounds. In particular, organocatalysis, which constitutes a complimentary extension to transition metal or enzymatic catalysis, has attracted much attention in recent years as a powerful, and environmentally friendly methodology.<sup>1</sup>

Most of the known organocatalytic synthesis are carried out through the accustomed classical stop-and-go approach, where intermediate isolation and purification steps are involved. In general, isolation and purification processes of intermediates are the most time- and cost-demanding and waste-producing manual operations in the traditional stop-and-go synthetic approach.<sup>2</sup>

In contrast to this approach stands the one-pot strategy, in which chemical transformations are performed without intermediary purification steps. One-pot reactions are therefore among the economically most attractive and sustainable methods in modern synthetic chemistry.<sup>2-5</sup>

Chiral 1,2-amino alcohols are versatile building blocks in the synthesis of biologically active target compounds and are subunits in many natural products and therapeutic agents, such as bronchodilators, neurotransmitters, sympathomimetic agents, and many others.<sup>6</sup> 1,2-Amino alcohols represent, therefore, attractive synthetic targets.

An example of two-step epoxide synthesis employing oxathiane as a chiral reagent and application of the isolated epoxides to preparation of bioactive 1,2-amino alcohols via aminolysis reaction with LiAl(NHR)<sub>4</sub>/NH<sub>2</sub>R was reported in 1995.<sup>7</sup>

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In 2011, we introduced a one-pot multi-step synthetic approach towards 1,2-amino alcohols employing oxazaborolidine organocatalysts, prepared in situ from  $\alpha$ -amino acids.<sup>8</sup> This new strategy we applied for the highly enantioselective synthesis of selected  $\beta$ -adrenergic blockers 1–3 (Fig. 1) via onepot sequential transformations: reduction of  $\alpha$ -halo ketones/ epoxide formation/aminolysis reaction.<sup>8</sup> Pharmaceuticals pronethalol  $(1)$ ,  $\int$ <sup>9</sup> nifenalol  $(2)$ <sup>10</sup> and dichloroisoproterenol  $(3)$ <sup>11</sup> are of great importance in the therapy of asthma, bronchitis and congestive heart failure.<sup>12</sup> **PAPER**<br> **CrossRation**<br> **CREATE:**<br> **CREATE:**<br>

Very recently, a new method for the preparation of  $(R)$ -pronethalol  $(1)$  and  $(R)$ -dichloroisoproterenol  $(3)$  in moderate yields and high enantioselectivities (41%, 96% ee and 60%, 98.6% ee, respectively) was developed, which involves acetylcyanation of prochiral aldehydes catalysed by a combination Ti-based Lewis acid catalyst and a biocatalyst employing a minor enantiomer recycling procedure, followed by a subsequent LAH or  $BH<sub>3</sub>$ reduction, respectively, and reductive amination using  $N$ aBH<sub>4</sub>/ acetone.<sup>13</sup>

Some practical shortcomings of the previously reported synthetic methodologies towards selected pharmaceuticals includes the involvement of multiple reaction steps, $7,8,13$  the requirement of isolation and purification of product intermediates<sup>7,13</sup> and the use of a relatively complex catalytic system (a combination of metal catalyst and biocatalyst)<sup>13</sup> and Li-based reagents (LiAl(NHR)<sub>4</sub>, LAH).<sup>7,13</sup> To match the basic needs of





Department of Chemistry and Pharmacy, Chair of Organic Chemistry I and Interdisciplinary Center for Molecular Materials (ICMM), University of Erlangen-Nürnberg, Henkestrasse 42, 91054 Erlangen, Germany. E-mail: svetlana.tsogoeva@fau.fe; Fax: +49-9131-85-26865 † Electronic supplementary information (ESI) available. CCDC 994083. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c4ra04011b

the pharmaceutical chemistry, the synthetic methods towards chiral drugs must be simple, straightforward and environmentally friendly. Thus, finding a shorter, facile and metal-free strategy providing a more convenient direct access to these  $chi$   $\beta$ -adrenergic antagonists under mild reaction conditions and without intermediary workup and purification steps, further reducing costs and waste, is very welcome.

Here we report a significant progress towards this goal with the development of a simple and practical organocatalytic epoxidation/aminolysis sequential one-pot route to the synthesis of  $\beta$ -adrenergic blockers (Fig. 2).

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Ar \ll \underset{\text{epoxidation}}{\underbrace{\text{organocatalytic}}}\Bigg[Ar \ll \hspace{-1.2cm}\sqrt{0}\Bigg] \; \underset{\text{aminolysis}}{\overset{\text{R}^1-NH_2}}\; \underset{\text{Ar}}{\underbrace{\hspace{1.2cm}\text{OH}}}\; H \ll \hspace{-1.2cm}\text{N} \ll \hspace{-1.2cm}\
$$

Fig. 2 Depiction of the designed short enantioselective synthesis of chiral 1,2-amino alcohols.



Fig. 3 Shi catalyst applied for epoxidation step.

Table 1 Optimisation of epoxidation reaction conditions

## Results and discussion

From a synthetic perspective, epoxides are particularly attractive and versatile starting compounds in organic synthesis and in the pharmaceutical industry.<sup>14,15</sup>

For realisation of the proposed synthetic route (Fig. 2), asymmetric epoxidation utilising organocatalysts has attracted our attention. Epoxidation of styrenes (selected here as a key reaction step) is still a challenging subject in the field of oxidation chemistry. An attractive metal-free method described for the epoxidation of terminal alkenes is the Shi epoxidation.<sup>16,17</sup> However, as far as we are aware, this method has never been applied to the synthesis of b-adrenergic blockers. We envisaged that employing versatile Shi epoxidation reaction as a key step to build up selected  $\beta$ -adrenergic blockers 1–3 in a onepot process would be of synthetic value. Paper<br>
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The Shi's N-substituted oxazolidinone ketone 4 (Fig. 3) has proven to be one of the most active types of fructose-derived asymmetric organocatalysts for epoxidation of unfunctionalised terminal alkenes.

Initially, we examined the epoxidation of 2-vinylnaphthalene in the presence commercially available N-substituted oxazolidinone ketone 4 using three conventional solvents (n-hexane, CH3CN and THF) different from DME (dimethoxyethane) originally reported for Shi epoxidation reaction.<sup>17</sup>

The laboratory work with toxic DME is known to imply several risks, for instance it may impair fertility and furthermore cause fetal harm.<sup>18</sup> Since the  $\beta$ -blockers 1–3 are administered as drugs it would be very desirable to find a comparable



<sup>a</sup> All reactions were carried out with Shi cat. 4 (0.15 equiv.), Oxone® (2.6–4.0 equiv.), and K<sub>2</sub>CO<sub>3</sub> (4 equiv., 0.84 M in  $4 \times 10^{-4}$  M aq. EDTA) and Bu<sub>4</sub>NHSO<sub>4</sub> (0.09 equiv.) in different solvents and buffer (0.1 M K<sub>2</sub>CO<sub>3</sub>-AcOH in  $4 \times 10^{-4}$  M aq. EDTA, pH = 9.3) unless otherwise stated.<br><sup>b</sup> Determined by <sup>1</sup>H-NMR. <sup>c</sup> Determined by chiral phase HPLC analysis and configurations were assigned by comparing HPLC elution order with known literature data. <sup>d</sup> Not determined. <sup>e</sup> The reaction was carried out at r.t.  $\overline{f}$  Oxone®: (2KHSO<sub>5</sub> KHSO<sub>4</sub> K<sub>2</sub>SO<sub>4</sub>).

Thus, improved conditions to those previously reported<sup>17</sup> for the Shi epoxidation of terminal alkenes, suitable for designed one-pot transformation (Fig. 1), were found. We have next applied this optimized method for the synthesis of epoxides 6 and 7, which are ideal precursors for  $(R)$ -nifenalol  $(2)$  and  $(R)$ dichloroisoproterenol (3) synthesis, respectively. After 6 h reaction time both epoxides were obtained with excellent to full conversion and enantiomeric ratios of 81 and 84%, respectively (entries 8 and 9).

The next aim was to combine this epoxidation reaction (without work up of the terminal epoxide) with the second aminolysis reaction step in a one-pot process (Table 2).

We started with the development of the one-pot process for (R)-pronethalol (1) synthesis. For the epoxidation reaction, Oxone® and  $K_2CO_3$  were added in a buffer solution dropwise simultaneously via two syringe pumps in order to ensure an ideal performance of the Shi catalyst 4 with a constant pH value during the epoxidation. After 6 h reaction time in THF we added 18 equiv. of isopropylamine to the reaction mixture and stirred further at 50  $\degree$ C for 18 h. We used 18 equiv. of isopropylamine, since our previous studies<sup>8</sup> demonstrated that an excess of isopropylamine is crucial for the aminolysis step for a complete consumption of epoxide.

 $(R)$ -pronethalol  $(1)$  was isolated in 55% yield after two sequential steps (entry 2). Enantiomeric ratios were reproducibly high (up to 90%, entries 1–5). By recrystallization, we

Table 2 One-pot two-step sequential synthesis of  $\beta$ -blockers 1-3

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	solvent without any loss of conversion and enantiomeric excess but at the same time with a less harmful effect.	Table 2 One-pot two-step sequential synthesis of $\beta$ -blockers 1-3	
	Hence, it was our motivation first to carry out preliminary solvent screening in the epoxidation reaction step (Table 1). The	Shi catalyst 4 Oxone@'	
	epoxidation of 2-vinylnaphthalene to epoxide 5 was performed	THF, 0 °C, 6 h	
	at 0 °C, beginning with 4.0 equiv. of Oxone® as an oxidizing agent. Both, the catalytic efficiency and asymmetric induction	Yield <sup>b</sup> (%) over two steps $Entrv^a$ Products	ee $^c$ (%)
	are strongly dependent on the solvents used. In $n$ -hexane and	$\mathbf{1}$ n.d	89 $(99)^d$
	acetonitrile, for instance, the reaction didn't work at all or only	$\boldsymbol{2}$ 55	81
	with a low conversion, respectively (entries 1 and 2). To our	$\mathbf{n} \cdot \mathbf{d}$ 3 n.d.f 4	90 $(99)^{d,e}$ 84
	delight, changing the reaction medium to THF, we observed a	n.d.' 5	87
	high conversion of 94% with an enantiomeric excess of 82%	6 <sup>g</sup> 44	84 $(99)^d$
	after only 6 hours of reaction time (entry 3). Furthermore, THF	$7^{g,h}$ 59	87
	offers another big advantage for the whole reaction system,	n.d.' 8 $\mathrm{n.d.}^f$ 9	84 82
	since previously, our group identified this solvent as the most	10 60	85 $(93)^d$
Article. Published on 16 July 2014. Downloaded on 10/31/2024 7:33:28 PM. This article is licensed under a Creative Commons Attribution 3.0 Unported Licence.	suitable for the following aminolysis reaction step. <sup>8</sup>	n.d.f 11	$>99^d$
	For further reaction optimization, the amounts of the oxidizing agent was reduced to 2.6 equiv. Carrying out the	78 12	83 $(99)^d$
	reaction for 6 h in THF at 0 °C the product was generated with		
	>99% conversion and 83% ee (entry 4, Table 1). To investigate	3	
	the influence of the temperature on the reaction outcome, we		
	carried out the epoxidation also at room temperature. This	$^a$ All reactions were carried out with Shi cat. 4 (0.15 equiv.), Oxone® (4.0)	
	experiment gave evidence that an elevated temperature affected	equiv.), and $K_2CO_3$ (4 equiv., 0.84 M in 4 $\times$ 10 <sup>-4</sup> M aq. EDTA) and Bu <sub>4</sub> NHSO <sub>4</sub> (0.09 equiv.) in THF and buffer (0.1 M K <sub>2</sub> CO <sub>3</sub> -AcOH in 4 $\times$	
	the conversion and enantioselectivity in a negative way (entry 5),	$10^{-4}$ M aq. EDTA, pH = 9.3) unless otherwise stated. <sup>b</sup> Obtained via	
	possibly caused by a decomposition of the used catalyst during	recrystallization. <sup>c</sup> Determined by chiral phase HPLC analysis and	
	the reaction course. <sup>19</sup> Among the different reaction times that	compared with authentic racemic material after isolation via preparative TLC. The absolute configuration was established by	
	were applied for the epoxidation, an extended reaction period	comparing HPLC elution order with known literature data. <sup>8</sup> d ee-value	
	(8 h, entry 6) didn't have any significant effect on the reaction	after a single recrystallization. <sup>e</sup> Single crystal for the X-ray diffraction analysis of the $(R)$ -pronethalol $(1)$ was obtained and the structure was	
	outcome, while a shortened period (4 h, entry 7) reduced the	unambiguously determined by X-ray crystallography (see the ESI <sup>†</sup> ). <sup>20</sup>	
	conversion to 90%. Thus, 6 h reaction time proved to be the	<sup>f</sup> Not determined. <sup>g</sup> DME was used as solvent. <sup>h</sup> Carried out at -10 °C.	
	most optimal for the epoxidation step.	<sup>l</sup> Oxone®: (2KHSO <sub>5</sub> · KHSO <sub>4</sub> · K <sub>2</sub> SO <sub>4</sub> ).	
Open Access	Thus, improved conditions to those previously reported <sup>17</sup> for		
	the Shi epoxidation of terminal alkenes, suitable for designed	further improved the ee-value of the product with respect to the	
	one-pot transformation (Fig. 1), were found. We have next	value determined directly from the crude reaction mixture.	
	applied this optimized method for the synthesis of epoxides 6		

 $^a$  All reactions were carried out with Shi cat. 4 (0.15 equiv.), Oxone® (4.0 equiv.), and K<sub>2</sub>CO<sub>3</sub> (4 equiv., 0.84 M in 4  $\times$  10<sup>-4</sup> M aq. EDTA) and Bu<sub>4</sub>NHSO<sub>4</sub> (0.09 equiv.) in THF and buffer (0.1 M K<sub>2</sub>CO<sub>3</sub>–AcOH in 4  $\times$  $10^{-4}$  M aq. EDTA,  $pH = 9.3$ ) unless otherwise stated. <sup>b</sup> Obtained *via* recrystallization. <sup>c</sup> Determined by chiral phase HPLC analysis and compared with authentic racemic material after isolation via preparative TLC. The absolute configuration was established by comparing HPLC elution order with known literature data.<sup>8</sup>  $d$  ee-value after a single recrystallization.  $e$  Single crystal for the X-ray diffraction analysis of the  $(R)$ -pronethalol  $(1)$  was obtained and the structure was unambiguously determined by X-ray crystallography (see the ESI†).<sup>21</sup> Not determined. <sup>*g*</sup> DME was used as solvent. <sup>*n*</sup> Carried out at -10 °C. *i* Oxone®: (2KHSO<sub>5</sub> · KHSO<sub>4</sub> · K<sub>2</sub>SO<sub>4</sub>).

further improved the ee-value of the product with respect to the value determined directly from the crude reaction mixture. Crystals were obtained in an almost enantiopure quality (up to 99% ee, entries 1 and 3) after single recrystallization.

In addition, we carried out the first step of the one-pot reaction under original conditions, applied by Shi et al. for the epoxidation reaction in DME at  $-10$  °C,<sup>17</sup> and were delighted to find that the catalyst 4 performed even better in THF at 0 °C than in DME at 0 °C or  $-10$  °C (entries 1-5 vs. entries 6 and 7). Thus, within our current research, we found that THF was a more suitable solvent than DME for the one-pot reaction system with respect to both the yield and the enantiomeric ratio.

Subsequently, we also applied the developed one-pot reaction route to the synthesis of two additional target compounds  $(R)$ -nifenalol  $(2)$  and  $(R)$ -dichloroisoproterenol  $(3)$ . The corresponding products 2 and 3 were obtained with 60% and 78% yields after two sequential steps, respectively (entries 10 and 12). Regarding the enantiocontrol of these reactions, we gained similar good results as compared to the  $(R)$ -pronethalol synthesis (entries 8–12 versus entries 1–5).

It is indicated, that the prepared  $\beta$ -blockers crystallize as conglomerates, since on recrystallization of 2 and 3, we again observed an enantioenrichment in the products as the obtained crystals had a signicantly higher ee-value (93% and 99%, respectively) as compared to the dissolved products in the crude reaction mixture (85% and 83%, entries 10 and 12, respectively).

## Conclusion

In summary, a new one-pot process providing a convenient twostep route for the formation of  $\beta$ -blockers:  $(R)$ -pronethalol  $(1)$ ,  $(R)$ -nifenalol  $(2)$  and  $(R)$ -dichloroisoproterenol  $(3)$ , from corresponding terminal alkenes was developed.

While the individual reactions selected for realisation of the one-pot synthesis (Shi-epoxidation and aminolysis reaction) are already known, the combination of these transformations in a one-pot two-step process leading to chiral drugs is unprecedented.

The use of commercially available Shi's N-substituted oxazolidinone ketone 4 as an organocatalyst for the epoxidation of terminal alkenes, followed by the aminolysis reaction with isopropylamine, and the mild reaction conditions used for both transformations, make this one-pot two-step process – up to the present – the most simple, shortest and environmentally friendly approach to obtain  $\beta$ -blockers.

## Experimental

#### General information

All reagents purchased from Sigma-Aldrich Co. or Acros Organics were used without further purification.  $n$ -Hexane, CH3CN, THF (HPLC grade) and DME was obtained from Fisher Scientific and used as received. Column chromatography was performed on Acros silica gel 60 Å and TLC on Macherey-Nagel ALUGram® SIL G/UV254 plates.  $^{1}$ H NMR and  $^{13}$ C NMR spectra were recorded on Bruker Avance 400 or Bruker Avance 300 NMR spectrometers using the solvent as internal standard. The enantiomeric excess of products was determined by chiral HPLC analysis in comparison with authentic racemic material. HPLC measurements were performed using Agilent Technologies 1200 Series equipment. Paper<br>
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restriction inster (63% and 83%, entries to and 15, respectively).<br>  $\mu_1$ , 7.31 (dd, 1H, J – 1.6 Hz, and 8.4 Hz), 4.33 (

#### General procedure for the asymmetric epoxidation

To a solution of the corresponding alkene (0.20 mmol, 1 equiv.) and Shi catalyst 4 (0.010 g, 0.030 mmol, 0.15 equiv.) in THF (3.0 mL) were added buffer (0.1 M K<sub>2</sub>CO<sub>3</sub>-AcOH in 4  $\times$  10<sup>-4</sup> M aq. EDTA, pH = 9.3, 2.0 mL) and Bu<sub>4</sub>NHSO<sub>4</sub> (0.0060 g, 0.018 mmol, 0.09 equiv.) with stirring. After the mixture was cooled to  $0^{\circ}$ C (bath temperature) via an ice bath, a solution of Oxone® (0.33 g, 0.53 mmol, 2.6 equiv.) in 4  $\times$  10<sup>-4</sup> M aq. EDTA (2.5 mL) and a solution of  $K_2CO_3$  (0.84 M in 4  $\times$  10<sup>-4</sup> M aq. EDTA, 2.5 mL) were added dropwise simultaneously within 6 h via syringe pumps. The reaction was quenched by dilution with DCM and extracted with DCM. The combined organic layers were dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , concentrated, and purified by flash chromatography (SiO2, petroleum ether–ethyl acetate) to give the corresponding epoxides as a colorless solid.

#### 2-(Naphthalen-2-yl) oxirane  $(5)^{21}$

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.82-7.89 (m, 4H), 7.49-7.45 (m, 2H), 7.31 (dd, 1H,  $J = 1.6$  Hz, and 8.4 Hz), 4.02 (dd, 1H,  $J = 2.8$ Hz, and 4.0 Hz), 3.21 (dd,  $1H, J = 4.0$  Hz, and 5.2 Hz), 2.90 (dd, 1H,  $J = 2.4$  Hz, and 5.6 Hz). Chiral HPLC (Chiralpak IC, nhexane–2-PrOH 99 : 1, flow rate 0.5 mL min<sup>-1</sup>,  $\lambda = 254$  nm):  $t_k$ 18.2 min (R), 21.5 min (S).

#### 2-(4-Nitrophenyl) oxirane  $(6)^{8,22}$

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 8.19 (d, 2H, J = 8.8 Hz), 7.43 (d,  $2H, J = 8.7$  Hz), 3.94 (dd, 1H,  $J = 2.4$  Hz, and 4.0 Hz), 3.21 (dd, 1H,  $J = 4.2$  Hz, and 5.5 Hz), 2.76 (dd, 1H,  $J = 2.4$  Hz, and 5.5 Hz). Chiral HPLC (Chiralpak IC,  $n$ -hexane-2-PrOH 95 : 5, flow rate 1 mL min<sup>-1</sup>,  $\lambda = 254$  nm):  $t_R$  13.9 min (R), 14.8 min (S).

#### $(3,4$ -Dichlorophenyl)oxirane  $(7)^{23}$

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.39 (d, 2H, J = 8.2 Hz), 7.33 (d,  $1H, J = 2.0$  Hz), 7.09 (dd,  $1H, J = 2.0$  Hz, and 8.3 Hz), 3.79 (dd, 1H,  $J = 2.6$  Hz, and 3.9 Hz), 3.12 (dd, 1H,  $J = 4.1$  Hz, and 5.4 Hz), 2.71 (dd,  $1H, J = 2.5$  Hz, and 5.6 Hz). Chiral HPLC (Chiralpak IC, *n*-hexane–2-PrOH 98 : 2, flow rate 1 mL min<sup>-1</sup>,  $\lambda = 280$  nm):  $t_k$ 6.1 min  $(R)$ , 7.2 min  $(S)$ .

#### General procedure for the one-pot two-step synthesis of  $\beta$ blockers

To a solution of the corresponding alkene (0.20 mmol, 1 equiv.) and Shi catalyst 4 (0.010 g, 0.030 mmol, 0.15 equiv.) in THF (3.0 mL) were added buffer (0.1 M K<sub>2</sub>CO<sub>3</sub>-AcOH in  $4 \times 10^{-4}$  M aq. EDTA, pH = 9.3, 2.0 mL) and  $Bu_4NHSO_4$  (0.0060 g, 0.018 mmol, 0.09 equiv.) with stirring. After the mixture was cooled to 0  $^{\circ}$ C (bath temperature) via an ice bath, a solution of Oxone®  $(0.33 \text{ g})$ , 0.53 mmol, 2.6 equiv.) in 4  $\times$  10<sup>-4</sup> M aq. EDTA (2.5 mL) and a solution of  $K_2CO_3$  (0.84 M in 4  $\times$  10<sup>-4</sup> M aq. EDTA, 2.5 mL) were added dropwise simultaneously within 6 h via syringe pumps. To this mixture isopropylamine (0.30 mL, 3.5 mmol, 18 equiv.) was added and the stirring was continued at 50  $\degree$ C for 18 hours. The organic solvent was evaporated and the remaining aqueous layer was extracted with ethyl acetate  $(2 \times 15 \text{ mL})$ . The combined organic layers were dried over  $Na<sub>2</sub>SO<sub>4</sub>$ . The solvent was removed under reduced pressure and the residue was recrystallized from ethyl acetate–n-hexane to give the corresponding  $\beta$ -blockers as colorless crystals.

#### $(R)$ -Pronethalol  $(1)^{8,24}$

White solid. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 7.80$ –7.83 (m, 4H), 7.43–7.47 (m, 3H), 4.81 (dd, 1H,  $J = 3.68$ , and 8.60 Hz), 3.01 (dd,  $1H, J = 3.75$  Hz, and 12.07 Hz), 2.83 (sept,  $1H, J = 6.26$  Hz), 2.72 (dd, 1H,  $J = 8.68$  Hz, and 12.07 Hz), 1.07 (dd, 6H,  $J = 1.85$  Hz, and 6.24 Hz). <sup>13</sup>C-NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 139.9, 132.8,$ 132.5, 127.7, 127.5, 127.2, 126.6, 125.3, 124.1, 123.6, 71.6, 54.1, 48.3, 22.7, 22.5. Chiral HPLC (Chiralpak IC, n-hexane–2-PrOH– ethanolamine 94.9 : 5 : 0.1, flow rate 1 mL min<sup>-1</sup>,  $\lambda = 280$  nm):  $t_R$  8.3 min  $(R)$ , 10.4 min  $(S)$ .

#### $(R)$ -Nifenalol  $(2)^{8,24}$

White solid. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 8.16$  (d, 2H,  $J = 8.4$ Hz), 7.51 (d, 2H,  $J = 8.5$  Hz), 4.82 (dd, 1H,  $J = 3.2$ , and 9.2 Hz), 2.96 (dd, 1H,  $J = 3.4$ , and 12.2 Hz), 2.40–2.64 (m, 1H), 2.60 (dd, 1H,  $J = 9.3$ , and 12.2 Hz), 1.11 (d, 3H,  $J = 6.7$  Hz), 1.10 (d, 3H,  $J =$ 6.7 Hz). <sup>13</sup>C-NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 149.9, 146.8, 126.0,$ 123.2, 70.4, 53.7, 48.4, 22.7, 22.4. MS (EI) m/z (%): 151 (5), 72 (100). Chiral HPLC (Chiralpak IC, n-hexane–2-PrOH–triethylamine 94.9 : 5 : 0.1, flow rate 0.5 mL  $\text{min}^{-1}$ ,  $\lambda = 280 \text{ nm}$ ):  $t_R$  25.1 min  $(R)$ , 26.5 min  $(S)$ .

#### $(R)$ -Dichloroisoproterenol  $(3)^{8,24}$

White solid. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.44 (d, 1H, J = 1.88 Hz), 7.37 (d, 1H,  $J = 8.26$  Hz), 7.15 (dd, 1H,  $J = 1.62$  Hz, and 8.28 Hz), 4.57 (dd, 1H,  $J = 3.61$  Hz, and 9.01 Hz), 2.86 (dd, 1H,  $J =$ 3.64 Hz, and 12.12 Hz), 2.77 (sept,  $1H, J = 6.29$  Hz), 2.54 (dd, 1H,  $J = 9.03$ , and 12.13 Hz), 1.04 (d, 6H,  $J = 6.28$  Hz). <sup>13</sup>C-NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 143.2, 132.4, 131.1, 130.2, 127.7, 125.1, 70.7,$ 54.3, 48.7, 23.1, 22.9. MS (EI) m/z: 230 (M–OH, 13%), 214 (M–Cl, 100). Chiral HPLC (Chiralpak IC, n-hexane–2-PrOH–ethanolamine 94.9 : 5 : 0.1, flow rate 0.5 mL  $\text{min}^{-1}$ ,  $\lambda = 280$  nm):  $t_R$  10.3 min  $(R)$ , 11.1 min  $(S)$ .

### Acknowledgements

We thank Dr Frank Hampel, Institute of Organic Chemistry I, for the X-Ray structure determination. S. B. T. is grateful to the Interdisciplinary Center for Molecular Materials (ICMM) and Erlangen Catalysis Resource Center (ECRC) for research support.

### Notes and references

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