

A novel antimicrobial agent reduces oxidative stress in cells†

Cite this: *RSC Advances*, 2013, 3, 7277

Received 6th February 2013,
Accepted 21st March 2013

DOI: 10.1039/c3ra40658j

www.rsc.org/advances

1,3,5-Tris[(1*E*)-2'-(4''-benzoic acid)vinyl]benzene] (Ramizol™) is a potent and non-toxic synthetic antimicrobial agent, and we now establish that it is also a potent inhibitor of reactive oxygen species (ROS) generation, with similar antioxidant activity to α -tocopherol (Vitamin E), which is a standard antioxidant drug. The results show that Ramizol™ decreases ROS to basal levels in PC12 cells and mouse and guinea pig myocytes, and demonstrate its protective nature from oxidative damage to the cells. The continuing challenge to discover antibiotics with no or few side effects makes Ramizol™ a very attractive candidate for the potential treatment of infections.

Introduction

An antioxidant is a molecule capable of slowing or preventing the oxidation of other molecules *via* the transfer of electrons. In general, oxidation reactions can produce free radicals which start chain reactions that damage cells. Antioxidants terminate such reactions by removing free radical intermediates and inhibit other oxidation reactions by being oxidized themselves.¹ Oxidative stress may induce cell damage, ultimately leading to cell death by apoptosis or necrosis. An increasing number of studies show that antioxidants can block cell death and may have therapeutic properties in animal models of diseases such as Alzheimer's, Parkinson's, hypertension, atherosclerosis, stroke, diabetes, cancer and inflammatory disorders.^{2,3} Therefore, in modern Western medicine, the balance between antioxidation and oxidation is believed to be critical for maintaining a healthy biology,^{4,5} and the status of protective mechanisms against oxidants is important. The antioxidants in humans reflect the dynamic balance between antioxidant defence and prooxidant conditions and they have

been suggested as a useful tool in estimating the risk of oxidative damage.^{6–8}

The symmetrical molecule 1,3,5-tris[(1*E*)-2'-(4''-benzoic acid)vinyl]benzene] (Ramizol™) (Fig. 1) has recently been established as a potent antimicrobial molecule with a minimum inhibitory concentration (MIC) against MRSA of 2 $\mu\text{g ml}^{-1}$.⁹ Ramizol™ shows no cytotoxicity in L929 cells (up to 500 $\mu\text{g ml}^{-1}$) and no haemolytic activity in sheep erythrocytes up to 50 $\mu\text{g ml}^{-1}$, at 25 times the MIC of MRSA. Moreover the AMES test establishes that there is no mutagenic activity due to Ramizol™ against the test organism *S. typhimurium* for concentrations up to 300 $\mu\text{g ml}^{-1}$.⁹ Its design as an antimicrobial compound was based on molecular docking experiments with the mechanosensitive channel of large conductance (MscL) protein.¹⁰ In the present study we investigate the antioxidant properties of Ramizol™ in different oxidative stress conditions in PC12 cells and mouse and guinea pig myocytes and discuss the scope for further strategic development of new therapies.

Materials and methods

Reagents

The reagents 1,3,5-tribromobenzene, $\text{Pd}_2(\text{dba})_3 \cdot \text{CHCl}_3$, *t*-Bu₃PHBF₄, Cy₂NMe and ethyl 4-vinylbenzoate (for the synthesis of Ramizol™), poly-L-lysine, menadione, EDTA, glutamate and D-

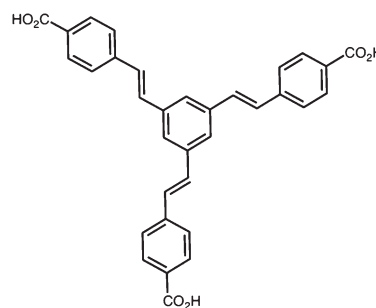


Fig. 1 Chemical structure of 1,3,5-Tris[(1*E*)-2'-(4''-benzoic acid)vinyl]benzene] (Ramizol™).

^aCentre for Strategic Nano-Fabrication, School of Chemistry and Biochemistry, The University of Western Australia, Crawley, WA 6009, Australia.

E-mail: ramiz.boulos@uwa.edu.au; Tel: 61 8 6488 4422

^bSchool of Anatomy, Physiology and Human Biology, The University of Western Australia M311, 35 Stirling Highway, Crawley, WA 6009, Australia

^cSchool of Chemical and Physical Sciences, Flinders University, Bedford Park, SA 5042, Australia. E-mail: colin.raston@flinders.edu.au

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c3ra40658j



α -tocopherol succinate were purchased from Sigma Aldrich, Australia. RPMI1640 cell culture media, fetal calf serum, horse serum, MEM non-essential amino acids, penicillin, streptomycin, PBS, DPBS, Hanks' balanced salt solution (HBSS), trypsin, trypsin-EDTA, sodium pyruvate, L-glutamine, cell dissociation buffer (enzyme free) and Live/Dead assay kit were purchased from Invitrogen Australia. The OxiSelect Intracellular ROS Assay Kit (Green Fluorescence) STA-342, was purchased from Cell Bio Labs., USA.

Synthesis of RamizolTM

The synthesis of RamizolTM followed the literature procedure with minor changes.¹¹ Ethyl acetate (20%) in hexane was used for monitoring the progress of the Heck cross coupling reaction between 1,3,5-tribromobenzene and ethyl 4-vinylbenzoate, using TLC and 20 : 80 ethyl acetate–dichloromethane for eluting the product. Purification of the final product from the saponification reaction involved column chromatography using a 20 : 80 methanol–tetrahydrofuran solvent system, followed by the addition of diethyl ether as an anti-solvent to wash the product.

Cell culture and treatment

Stock cultures of PC12 cells,¹² purchased from ATCC, were grown routinely in 75 cm² tissue culture flasks in RPMI 1640 supplemented with 10% (v/v) heat-inactivated horse serum, and 5% (v/v) heat-inactivated fetal calf serum, to which penicillin (50 U ml⁻¹) and streptomycin (50 μ g ml⁻¹) were added. Cultures were maintained at 37 °C in a humidified incubator containing 95% air and 5% CO₂. The cells were passed twice a week. One day prior to experimentation, the PC12 cells were plated on poly-L-lysine (10 μ g mL⁻¹) coated at a density of 10 000 cells per well in 96 well plates and 2 \times 10⁵ cells per well in 6 well plates. RamizolTM was prepared in different concentrations (3, 1, 0.3, 0.1, 0.01, 0.01 mg mL⁻¹) using RPMI media and then the PC12 cells were treated with this preparation for 24 h for cytotoxicity tests. α -Tocopherol (A-T) is the main vitamin E component used as a standard drug in ROS experiments.

Cell viability test

In order to assess the cytotoxic effect of RamizolTM on PC12 cells, we investigated the viability of PC12 cells incubated with a series of concentrations of RamizolTM from 3 to 0.001 mg mL⁻¹ for 24 h. After 24 h the RamizolTM was removed and the cells were washed 3 times with PBS after which 1 μ M calcein-AM and 2 μ M Ethidium homodimer (Live and dead assay kit) were added and the cells then incubated at 37 °C in 5% CO₂ for 30 min. Images were taken at 4 locations in each well for three replicates of each concentration using a Nikon Inverted Fluorescence microscope.

Effect of RamizolTM on intracellular reactive oxygen species (ROS) generation, in PC12 cells

Intracellular accumulation of hydrogen peroxide was determined with DCFH-DA. This non-fluorescent compound accumulates within cells upon deacetylation. DCFH then reacts with the reactive oxygen species to form fluorescent dichlorofluorescein (DCF). PC12 cells were plated in 96-well black and clear bottom plates and grown for 24 h before the addition of 1X DCFH-DA,

incubation for 60 min at 37 °C, and then treatment with RamizolTM at 0.1 mg mL⁻¹, A-T (10 μ M) and stressors 1 mM H₂O₂, 500 μ M menadione and 100 mM glutamate, respectively for 60 min. Cells were then washed 3 times with Hank's balanced salt solution (HBSS without phenol red) at room temperature. Cellular fluorescence was measured using an Enspire multimode plate reader using an excitation wavelength of 480 nm and emission wavelength of 530 nm (Software version 4.1.) (Perkin Elmer, USA). Results were expressed as Relative Fluorescence Units (RFU).

Effect of RamizolTM on H₂O₂ induced increase in superoxide production in adult mouse and guinea pig cardiac myocytes

Superoxide generation was assessed in individual intact adult mouse and guinea pig cardiac myocytes. Animals were anaesthetised using an intraperitoneal injection of pentobarbitone sodium (240 mg kg⁻¹) prior to the excision of the heart as approved by The Animal Ethics Committee of The University of Western Australia, in accordance with the Australian Code of Practice for the Care and Use of Animals for Scientific Purposes (NH&MRC, 7th Edition, 2004). Cells were isolated as previously described.¹³ Superoxide generation was assessed in the myocytes before and after transient exposure to 30 μ mol L⁻¹ H₂O₂ using the fluorescent indicator dihydroethidium (DHE, 5 μ M, 515–560 nm ex filter, 590 long pass em) at 37 °C as previously described.¹⁴ Transient exposure to 30 μ mol L⁻¹ H₂O₂ was achieved through the extracellular application of 30 μ mol L⁻¹ H₂O₂ for 5 min followed by 10 μ mol mL⁻¹ catalase to degrade the H₂O₂. The fluorescent signal was measured on a Hamamatsu Orca ER digital camera attached to an inverted Nikon TE2000-U microscope. Fluorescent images were taken at 1 min intervals with 200 ms exposure. Metamorph 6.3 was used to quantify the signal by manually tracing individual myocytes. An equivalent region not containing cells was used as background and was subtracted. Fluorescence was reported as a ratio of the slope of the signal measured at 16–30 min (treatment) over 1–15 min (basal). Basal fluorescence was assigned a value of 1.0.

Statistical analysis

Data were expressed as means \pm SEM of triplicate determinations from the independent experiments. The statistical significance of the data was assessed by standard analysis of variance (ANOVA) techniques followed by a Bonferroni test using Stat view version 5.0.1 software and *p* values \leq 0.05 were considered statistically significant.

Results

Cell viability test

To assess the cytotoxic effect of RamizolTM, PC12 cells were treated with RamizolTM at different concentrations ranging from 0.001 to 3 mg mL⁻¹ for 24 h, Fig. 2. The control group received only RPMI media and was analyzed using a live and dead assay kit. Fig. 3 represents the cell viability test for RamizolTM. The concentration at which the cell viability reduced by approximately 50% (IC₅₀ value) was established at 3 mg mL⁻¹ (47.38 \pm 6.2%) and the concentration used for the treatment of cells for all assays was found to be safer and protective, having 97.16 \pm 2.3% viable cells.



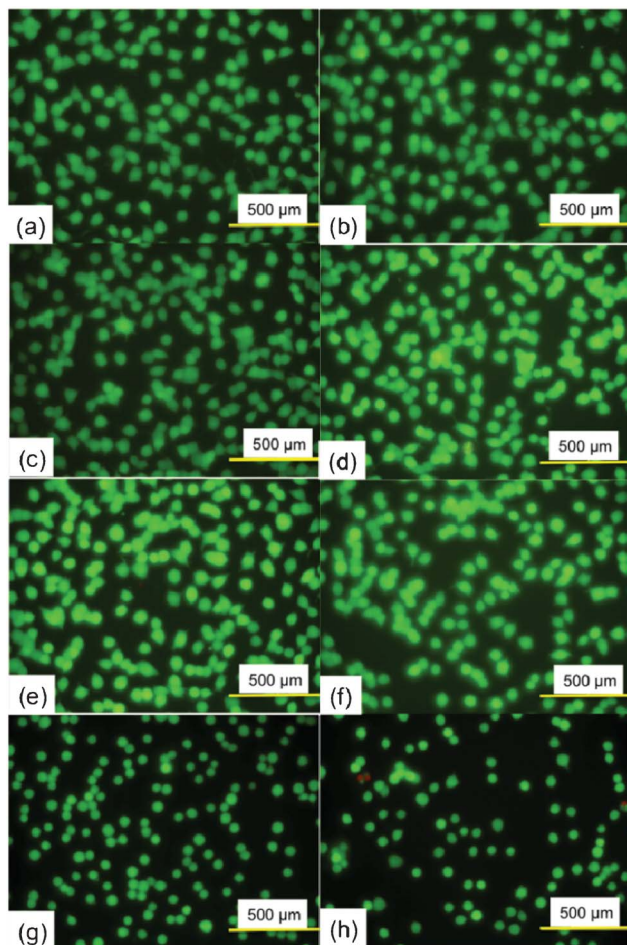


Fig. 2 Fluorescence images of PC12 cells in the presence of varying concentrations of RamizolTM. (a) and (b) 0 mg mL⁻¹, (c) 0.001 mg mL⁻¹ (d) 0.01 mg mL⁻¹ (e) 0.1 mg mL⁻¹ (f) 0.3 mg mL⁻¹ (g) 1 mg mL⁻¹ and (h) 3 mg mL⁻¹.

Effect of RamizolTM on intracellular ROS generation, in PC12 cells

The intracellular production of ROS was measured using a fluorescent probe, DCFDA, *via* a multimode plate reader. ROS production increased 10 fold after treatment with H₂O₂, menadione and glutamate treatment, Fig. 4, which was significantly ($p \leq 0.05$) different from the no treatment group. The group pretreated with RamizolTM showed a two-fold increase in ROS

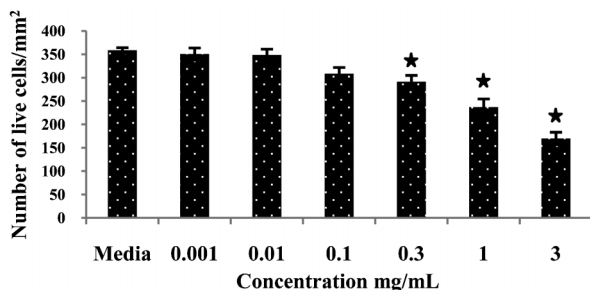


Fig. 3 Cytotoxic effect of a compound RamizolTM on PC12 cells (Live/Dead Assay).⁵

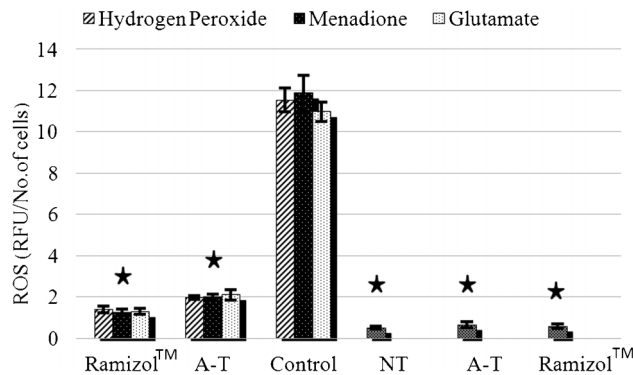


Fig. 4 Effect of RamizolTM on intracellular ROS generation in PC12 cells. § RamizolTM (3 columns): stressed PC-12 cells treated with RamizolTM at 0.1 mg mL⁻¹ concentration, A-T (3 columns): stressed PC-12 cells treated with α -tocopherol at 0.01 mg mL⁻¹ concentration, Control: control group treated with 3 different stressors without treatment with RamizolTM, NT (no treatment): PC-12 cells without oxidants and antioxidants treatment, A-T (1 column): PC-12 cells treated with α -tocopherol and RamizolTM (1 column): PC-12 cells treated with RamizolTM.

production and with α -tocopherol a three-fold increase, which was significantly different from the no treatment and control groups. There was no significant increase in the ROS generation observed in the groups treated only with RamizolTM and α -tocopherol, and they were significantly different ($p \leq 0.05$) from the control group. The cells pre-treated with RamizolTM showed a similar trend to the α -tocopherol treated group.

Effect of RamizolTM on H₂O₂ induced increase in superoxide production in adult mouse and guinea pig cardiac myocytes

In cardiac tissue, persistent increases in ROS are associated with pathological remodeling and myocardial dysfunction.¹⁵⁻¹⁷ There is good evidence that increased ROS production following exposure of cardiac myocytes to sub-lethal levels of ROS contributes to the development of cardiac hypertrophy.¹⁸⁻²¹ We examined the effect of the compound RamizolTM on H₂O₂ induced increase in superoxide production in individual adult mouse and guinea pig cardiac myocytes. Consistent with previous findings, transient exposure of mouse and guinea pig cardiac myocytes to a sub-lethal dose of H₂O₂ (30 μ mol L⁻¹) resulted in a significant increase in superoxide production ($n = 4$, $p < 0.05$), Fig. 5.¹⁴ Exposure of cardiac myocytes to 2, 10 or 20 μ g mL⁻¹ RamizolTM did not significantly alter this response. However, exposure of cardiac myocytes to 30 or 50 μ g mL⁻¹ RamizolTM completely abolished the increase in superoxide production in response to H₂O₂ exposure ($n = 4$ and $n = 5$ respectively, $p < 0.05$).

Discussion

Antioxidants are free-radical scavengers which can provide protection to living organisms from damage caused by uncontrolled production of reactive oxygen species.²² The synthetic antioxidants used most commonly today are butylated hydroxyanisole (BHA), butylated hydroxytoluene (BHT), propylgallate (PG) and butylated hydroquinone.^{23,24}



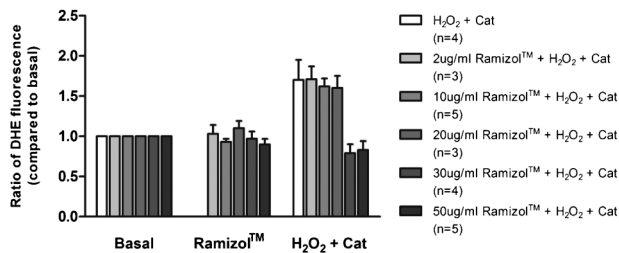


Fig. 5 Concentration-dependent effect of Ramizol™ on hydrogen peroxide-induced increase in superoxide generation in cardiac myocytes (n = number of cardiac myocytes).

ROS evaluation is used as an index to quantify the overall oxidative stress in cells.²⁵ The elevated ROS generation was observed when the cells were treated with 3 stressors independent of its mechanism of oxidation. But the intracellular ROS generation, Fig. 2, was reduced considerably when the cells were treated with Ramizol™, which was concordant with previous reports of effective ROS inhibition by hydroxytyrosol (HTy) and hydroxytyrosol acetate (HTy-A).²⁶

In cardiac tissue, persistent increases in ROS are associated with pathological remodeling and myocardial dysfunction.^{15–17} Increases in ROS beyond 100 $\mu\text{mol l}^{-1}$ (extracellular) are associated with mitochondrial damage and damage to key macromolecules such as DNA, proteins and lipids.^{15,27–29} This ultimately leads to myocyte loss and heart failure. Sub-lethal increases in ROS (1–100 $\mu\text{mol l}^{-1}$ extracellular) can result in the activation of a number of calcium-dependent signalling pathways that have been associated with pathological growth and progression towards cardiac hypertrophy in the absence of cell death.^{15,18–21,27,30,31} In myocytes, this is associated with increased protein synthesis in the absence of apoptosis or necrosis,^{18–21} and a small but significant increase in cell area 48 h following the initial exposure to H₂O₂.²¹

We tested the effect of Ramizol™ on ventricular myocytes isolated from mouse and guinea-pig hearts to determine whether the compound was effective after an oxidative stress challenge in native cells. Consistent with previous findings, we find that transient exposure to 30 $\mu\text{mol l}^{-1}$ H₂O₂ results in a significant increase in superoxide production in mouse and guinea pig cardiac myocytes.¹⁴ Exposure to concentrations of Ramizol™ within the range of 30–50 $\mu\text{g mL}^{-1}$ completely attenuated the increase in superoxide production in response to transient exposure of mouse and guinea pig cardiac myocytes to 30 $\mu\text{mol l}^{-1}$ H₂O₂.

The electrochemistry of Ramizol™ was studied at a buffered pH of 7 using phosphate with KNO₃ as the electrolyte. No redox peaks were observed between –0.4 V and 0.7 V vs. 3.5 M KCl Ag–AgCl (Fig. S1, ESI†). This range was extended to between –1.2 V and 0.7 V vs. 3.5 M KCl Ag–AgCl using a pH of 12 at 0.1 M phosphate solution with no observable peaks that could not be excluded from the electrodes' background. This potential range includes the electrochemical behaviour of the majority of biological redox active compounds such as vitamin B, C and E.^{32,33} The lack of redox peaks indicates that the antioxidant

behaviour of Ramizol™ either follows an auto-oxidation type mechanism, as has been observed in unsaturated fats, or a secondary role, such as that observed in saturated fats.^{34,35} Given the presence of carboxylic acid moieties and unsaturation in Ramizol™, auto-oxidation is a more plausible mechanism for the antioxidant behaviour of the drug but this is yet to be confirmed and will be a topic in future work.

We have shown that the novel and non-toxic antimicrobial, Ramizol™, is also a potent antioxidant. These findings make the drug an attractive candidate for fighting infections in patients with an on-going oxidative stress disorder or a compromised immune system.

Conclusion

Our study suggests that Ramizol™ is a potent synthetic compound that exhibits effective inhibition of ROS generation compared to α -tocopherol, which is a standard antioxidant drug. The results show that Ramizol™ decreases ROS to basal levels in PC12 cells and mouse and guinea pig myocytes, and demonstrate its protective nature from oxidative damage to the cells. The continual challenge to discover antibiotics with no or few side effects makes Ramizol™ a very attractive candidate for the potential treatment of infections.

Acknowledgements

The effect of Ramizol™ on adult mouse and guinea pig cardiac myocytes study was supported by grants from the National Health and Medical Research Council of Australia and Australian Research Council. Livia Hool is an Australian Research Council Future Fellow and Honorary National Health and Medical Research Council Senior Research Fellow. Helena Viola is a recipient of a National Heart Foundation of Australia Postdoctoral Fellowship.

Notes and references

† The authors would kindly like to thank Dr Adrian Scaffidi for the synthesis of Ramizol™.
 § Values are expressed as mean \pm SEM of three determinations.
 * indicates significant difference ($p < 0.05$) from the control group.

- 1 H. Sies, *Exp. Physiol.*, 1997, **82**, 291–295.
- 2 E. Esposito, D. Rattilio, V. Di Matteo, C. Di Glulia, M. Cacchio and S. Algeri, *Neurobiol. Aging*, 2002, **23**, 719–735.
- 3 C. Behl, *Prog. Neurobiol.*, 1999, **57**, 301–323.
- 4 T. Finkel, *Nature*, 2000, **408**, 239–248.
- 5 K. J. A. Davies, *IUBMB Life*, 2000, **50**.
- 6 K. Nose, *Biol. Pharm. Bull.*, 2000, **23**, 897–903.
- 7 A. M. Papas, *Lipids*, 1996, **31**, 77–82.
- 8 M. C. Polidori, W. Stahl, O. Eichler, I. Njestrroj and H. Sies, *Free Radical Biol. Med.*, 2001, **30**, 456–462.
- 9 *US Patent Appl.*, No. 13/517,455.
- 10 R. A. Boulos, PhD Thesis, *Chemistry*, University of Western Australia, Crawley, 2011.
- 11 N. A. Lengkeek, R. A. Boulos, A. J. McKinley, T. V. Riley, B. Martinac and S. G. Stewart, *Aust. J. Chem.*, 2011, **64**, 316–323.
- 12 L. A. Greene and A. S. Tischler, *Proc. Natl. Acad. Sci. U. S. A.*, 1976, **73**, 2424–2428.



- 13 L. C. Hool, *Circ. Res.*, 2000, **87**, 1164–1171.
- 14 H. M. Viola, *et al.*, *Circ. Res.*, 2007, **100**, 1036–1044.
- 15 A. Sabri, H. H. Hughie and P. A. Lucchesi, *Antioxid. Redox Signaling*, 2003, **5**, 731–740.
- 16 R. Bolli, *Basic Res. Cardiol.*, 1998, **93**, 156–162.
- 17 D. J. Grieve, *et al.*, *Heart, Lung Circ.*, 2004, **13**, 132–138.
- 18 S. H. Kwon, *J. Mol. Cell. Cardiol.*, 2003, **35**, 615–621.
- 19 Q. M. Chen, *et al.*, *Arch. Biochem. Biophys.*, 2000, **373**, 242–248.
- 20 V. C. Tu, *et al.*, *J. Pharmacol. Exp. Ther.*, 2002, **300**, 1101–1110.
- 21 V. Seenarain, *et al.*, *J. Proteome Res.*, 2010, **9**, 1985–1994.
- 22 S. Ghosal, V. K. Tripathi and S. Chaeruhann, *Indian J. Chem.*, 1996, **35**, 941–948.
- 23 H. P. Wichi, *Food Chem. Toxicol.*, 1988, **26**, 723–727.
- 24 E. R. Sherwin, A. L. Branch, P. M. Davidson and S. Salminen, *Food Additives*, Marcel Dekker Inc, New York, 1990.
- 25 B. Queguineur, L. Goya and S. Ramos, *et al.*, *Food Chem. Toxicol.*, 2012, **50**, 2886–2893.
- 26 G. Pereira-Caro, B. Sarria and A. Madrona, *et al.*, *J. Agric. Food Chem.*, 2011, **59**, 5964–5976.
- 27 J. R. Stone and S. Yang, *Antioxid. Redox Signaling*, 2006, **8**, 243–270.
- 28 D. B. Sawyer, *et al.*, *J. Mol. Cell. Cardiol.*, 2002, **34**, 379–388.
- 29 F. J. Giordano, *J. Clin. Invest.*, 2005, **115**, 500–508.
- 30 N. Frey and E. N. Olsen, *Annu. Rev. Physiol.*, 2003, **65**, 45–79.
- 31 V. P. van Empel and L. J. De Windt, *Cardiovasc. Res.*, 2004, **63**, 487–499.
- 32 W. W. Yao, H. M. Peng and R. D. Webster, *J. Phys. Chem. C*, 2009, **113**, 21805–21814.
- 33 H. Y. Gu, A. M. Yu and H. Y. Chen, *Anal. Lett.*, 2001, **34**, 2361–2374.
- 34 I. P. Freeman, in *Ullmann's Encyclopedia of Industrial Chemistry*, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, 2000.
- 35 G. E. Henry, R. A. Momin, M. G. Nair and D. L. Dewitt, *J. Agric. Food Chem.*, 2002, **50**, 2231–2234.

