The Raman spectrum of isolated water clusters†

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Cold water oligomers \((\text{H}_2\text{O})_n\) and \((\text{D}_2\text{O})_n\) with \(n = 2–5\) are assigned in spontaneous Raman scattering spectra of seeded rare gas expansions for the first time. Comparison with infrared spectra provides direct experimental insights into the hydrogen bond-mediated excitonic OH oscillator coupling, which is responsible for ultrafast energy transfer between water molecules, usually suppressed by isotopic dilution in femtosecond experiments for the condensed phase. The experimental coupling constants are compared to those in state-of-the-art full-dimensional water potential energy hypersurfaces, leaving room for improvement in the description of the coupled dynamics in water. Evidence for intensified Fermi resonance between OH stretching and OH bending motion beyond water trimers is collected.

1 Introduction

The vibrational dynamics of water is of fundamental interest.\(^1\) It is mediated by a network of intermolecular hydrogen bonds, which can be introduced step by step in the popular cluster approach.\(^2,3\) While matrix isolation\(^4,5\) and chromophore labeling\(^6,7\) have proven useful in this context, the study of unperturbed, isolated water clusters is particularly attractive due to the close contact with theory\(^8\) and the large amplitude motion in these systems. Direct infrared\(^2,9–11\) and microwave\(^12\) studies have become available in recent years and have provided detailed information on nuclear quantum effects and electronic cooperativity, in particular when combined with theoretical predictions. Although the Raman selection rules provide essential complementary information on concerted motions, only a pioneering coherent anti-Stokes Raman spectroscopy (CARS) investigation of isolated water clusters has so far been published\(^13\) and discussed controversially.\(^14,15\) The nonlinearity of the CARS experiment rendered a firm assignment of cluster sizes difficult. Therefore, the characteristic coupling between neighboring OH oscillators in water assemblies, which controls energy flow after local excitation, has remained largely in the dark.

This is where the present contribution sets in. By generating isolated, internally cold, small water clusters in slit jet expansions of an atomic gas into vacuum and probing them by an intense visible laser, sufficient numbers of spontaneously Raman-scattered photons are generated to detect OH stretching modes which show negligible infrared intensity and have thus remained unassigned in the gas phase to date. The strong hydrogen bond cooperativity ensures a sufficient spread of the signals on the wavenumber axis for a reliable cluster size separation and size assignment based on stagnation pressure, nozzle temperature, polarization dependence\(^16\) and water concentration dependence up to at least five water units. Further support for the vibrational and size assignment comes from equivalent experiments with \(\text{D}_2\text{O}\). This compensates for the lack of rigorous size selection.\(^6,17\) The combination with corresponding IR data\(^10,18\) then leads to an effective approach to the oscillator coupling within a simple framework of Hückel-like coupling matrices,\(^2\) as exemplified in the case of cyclic methanol clusters.\(^19\) The energy splitting of the degenerate monomer oscillators into symmetric and asymmetric linear combinations in a cyclic cluster is caused by coupling through space or through the connecting hydrogen bonds.

These coupled modes are observable as a mode pattern in the frequency domain,\(^20\) but their importance is better visualized in the time domain. Excitation of a local OH stretching mode by a suitable ultrashort laser pulse would lead to a rapid delocalization of the excitation energy among the coupled oscillators as a function of time. In a finite cluster and in the absence of other redistribution mechanisms, this is a periodic process which ultimately leads to a refocussing of the energy in the original oscillator. In liquid or solid water, the energy is dissipated irreversibly. Although the detailed quantum dynamics can be complex, there is a reciprocal relationship between the magnitude of the excitonic splitting and the lifetime of the localized state. The larger the spectral splitting, the faster the energy dissipation. Therefore, the size of the coupling constant between OH stretching modes in a water cluster is of

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prime importance for the femtosecond dynamics of these high frequency oscillators.\textsuperscript{16,21,22} Because the ultrafast exciton dynamics among degenerate oscillators leads to rapid anisotropy decay, most femtosecond experiments and simulations on liquid water are carried out in isotopic dilution to focus on slower processes.\textsuperscript{1,23,24} In liquid water, the effective excitonic coupling constant has been estimated\textsuperscript{25} to be around 12 cm\textsuperscript{-1}.

In cold ring-like clusters, it is modelled to be of the same order of magnitude for the bonded OH stretch mode.\textsuperscript{26} This is significantly smaller than in the related alcohol clusters\textsuperscript{19} (20–30 cm\textsuperscript{-1}), which feature the same hydrogen bonded ring topology.\textsuperscript{27} We will argue in this contribution that the experimental coupling constants in ring-like water clusters are indeed larger than analytically modelled and more in line with those observed in methanol clusters. The associated concerted stretching modes of water ring clusters are particularly interesting and important because they represent a low energy window of the potential energy hypersurface where water molecules lose their integrity and start to undergo isotope exchange.\textsuperscript{28} These modes are also predicted to have exceptionally fast IVR rates.\textsuperscript{29}

One important aspect has to be considered in the OH stretch coupling of small water clusters. In isolated water molecules, the OH bending overtone is located more than 500 cm\textsuperscript{-1} below the symmetric stretching fundamental\textsuperscript{30} but it comes into resonance for cooperatively weakened OH bonds. As we will argue based on Raman spectroscopic evidence, this hydrogen bond-tuned Fermi resonance becomes increasingly important when moving from the trimer to the pentamer, opening up another energy flow channel in particular for symmetric and thus Raman-active excitation. The corresponding normal modes are visualized for the pentamer case in Fig. 1.

While the focus of this contribution is on cyclic trimers, tetramers and pentamers with their low microwave visibility\textsuperscript{12} but well-characterized low-frequency dynamics,\textsuperscript{2,31} we also present the first Raman spectra for isolated water dimers, for which some transitions have previously been observed in He nanodroplets.\textsuperscript{32} Finally, we provide circumstantial evidence for the topologically versatile hexamer.\textsuperscript{2,12,33–36}

After a brief description of the experimental setup, the spontaneous Raman spectra will be presented and assigned. The discussion of excitonic patterns in the cyclic clusters will start with the most clearcut tetramer case and extend to the trimer and pentamer, followed by preliminary assignments of non-cyclic hexamer bands.

2 Experimental setup

Gas mixtures of water with He, Ne/He or Ar/He were expanded \textit{via} a 4 x 0.15 mm\textsuperscript{2} or 8 x 0.05 mm\textsuperscript{2} heatable slit nozzle into an aluminium chamber, pumped using two Roots pumps (500 m\textsuperscript{3} h\textsuperscript{-1} and 250 m\textsuperscript{3} h\textsuperscript{-1}) and one rotary vane pump (100 m\textsuperscript{3} h\textsuperscript{-1}). The beam of a frequency doubled cw NdYVO\textsubscript{4}-Laser (Coherent Verdi V18, 18 W, $\lambda=532$ nm) was focused onto the expansion at 2 mm distance from the nozzle exit. The scattered light was collected perpendicularly to the expansion and to the incident laser beam and collimated using a fast camera lens (50 mm Nikon, f/1.2).

It was then focused onto the entrance slit of the monochromator (McPherson Model 2051, f/8.6, $f=1000$ mm, grating 1200 grooves per mm or McPherson Model 205f, f/3.2, $f=500$ mm, grating 600 grooves per mm) using an achromatic planoconvex lens (Edmund Optics, $\psi=50$ mm, $f$ for McPherson Model 2051 and Edmund Optics, $\psi=50$ mm, $f/4$ for McPherson Model 205f). Suppression of the Rayleigh scattered light was achieved using a Raman edge filter (L.O.T., $\psi=25$ mm, OD 6.0, T > 90\%, 535.4–1200 nm). A back-illuminated CCD camera (PI Acton, Spec-10: 400 R/LN, 1340 x 400 pixels, cooled by liquid N\textsubscript{2}) served for detection in a vertical binning mode.\textsuperscript{19} Count rates refer to the full 400-pixel columns. The wavelength calibration of the spectra was carried out using the lines of a Ne I emission light source and H\textsubscript{2}O monomer Raman transitions.\textsuperscript{30,37} Cosmic ray signals were removed by the comparison of block-averaged spectra. Polarization experiments were performed by rotating the laser beam \textit{via} a $\lambda/2$-plate.

The Stokes scattering was probed from 80 to 8000 cm\textsuperscript{-1}, but here we concentrate on the by far strongest\textsuperscript{38} Raman cluster signals between 3000 and 4000 cm\textsuperscript{-1}, in the region of the OH stretching fundamentals and the OH bending overtones.

3 Results and discussion

3.1 Spectral survey

In supersonic jet expansions, different molecular degrees of freedom may have different effective temperatures or even non-thermal distributions. We obtained monomer rotational temperatures of 30–50 K based on the relative intensities of Stokes and anti-Stokes transitions (see ESI\textsuperscript{\dagger} for details). These are in qualitative agreement with recent results for cluster-free expansions of water in helium, considering the large differences in expansion conditions.\textsuperscript{17} Vibrational temperatures are probably higher and lead to relatively broad band contours with occasional hot band features, marked with * in the following spectra. The temperatures can be lowered by variation of the stagnation conditions and carrier gas.

Fig. 2 and 3 contain a collection of Raman supersonic jet spectra for H\textsubscript{2}O and D\textsubscript{2}O expansions in different carrier gases and under varying conditions (described in detail in the ESI\textsuperscript{\dagger}). Their spectral windows have been matched by D\textsubscript{2}O wavenumber

![Fig. 1 The in-phase hydrogen-bonded OH stretching fundamental is closely resonant with the overtone of the highest frequency bending mode in the water pentamer (and less so in the tetramer).](image-url)
axis stretching ($\times 1.508$) and monomer symmetric stretch band center alignment. The correspondence between the isotopologues is indeed very close. Studies of the carrier gas dependence and of the depolarization ratio are provided in the ESI.$^\dagger$ Table 1 provides peak positions and explains the labels of the Raman spectral features. Also shown in the upper right traces and in Table 1 are corresponding FTIR supersonic jet spectra, which agree with earlier laser spectroscopy observations.$^{18,39}$

The high degree of complementarity for trimers (3b), tetr democr (4b) and pentamers (5b) is a consequence of their ring topology.

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**Fig. 2** Raman and IR$^{10}$ spectra of mixed H$_2$O/rare gas expansions. Raman spectra were scaled to the same monomer scattering intensity at the point of measurement in the respective blocks. The labels are defined in the text and in Table 1. Detailed information on the measurement conditions, such as nozzle temperature $T_n$, saturator temperature $T_s$, and stagnation pressure $p_s$ can be found in Table SI in the ESI.$^\dagger$ Lower part: Raman spectra recorded using a 0.5 m monochromator. Basically, the clustering extent increases from bottom to top. Upper part: similar conditions but recorded using a 1 m monochromator. Again, the clustering extent increases from bottom to top.

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**Fig. 3** Like Fig. 2, but for D$_2$O/rare gas expansions, also showing trimer to pentamer structures. For detailed information see Table SII in the ESI.$^\dagger$
The Raman spectra are dominated by the sharp symmetric OH stretching transitions of the monomer, some of which are labeled (1). Between the symmetric and antisymmetric water monomer bands, a structured scattering signal from dangling (free) OH groups is found. Contributions from dimers (2f), trimers (3f, >3f), larger (>3f) and significantly larger (≫3f) clusters can be partially separated, but their discussion will be postponed. They are also summarized in Table 1 for both light and heavy water.

A general feature of the warmer Raman spectra is the asymmetric slope of the hydrogen-bonded stretching bands due to thermal excitation. This reflects the sensitivity of the OH stretching wavenumber to the strength of the hydrogen bond, so important in fs bulk water experiments and in the room temperature detection of water clusters. The colder spectra using heavier carrier gases bonds arranged on the sides of a square. Each OH group has the same zeroth order wavenumber and couples directly to its hydrogen bond acceptor partner and either along two hydrogen bonds or across the diagonal to its next-nearest neighbor. The direct coupling is described by $W_4$ and the weaker diagonal coupling by $W_2$. A Hückel-like analysis of the combined action of all couplings leads to the determinant equation

$$\begin{vmatrix} x & W_1 & W_2 & W_1 \\ W_1 & x & W_1 & W_2 \\ W_2 & W_1 & x & W_1 \\ W_1 & W_2 & W_1 & x \end{vmatrix} = 0$$

The solutions of which are

$$x_1 = -2W_1 - W_2 \quad (A)$$

$$x_{2,3} = W_2 \quad (E)$$

$$x_4 = 2W_1 - W_2 \quad (B)$$

The symmetry labels A, E, B derive from the $S_4$ point group. The degenerate pair of E transitions, which is predominantly IR-active, is thus separated by $W_2$ from the center of the coupling pattern, whereas the predominantly Raman-active A transition is below the center, by $2W_1 + W_2$. This A level was recently predicted to have an intracluster energy redistribution rate of 0.4 ps. The observed width of the Raman transition of 10 cm$^{-1}$ is qualitatively still compatible with this rate in the limit of a dense manifold of coupling states, but actually more narrow than the corresponding IR mode. The B-transition is expected to have a weak intensity in both types of spectra. The splitting between the IR-active and dominant Raman active bands (E/A) is thus $2W_1 + 2W_2$. Between the weaker Raman-active and the IR-active bands (B/E) it is $2W_1 - 2W_2$, and between the two Raman active bands (B/A) $4W_1$. As the assignment of the second Raman active band is less certain, we define a composite coupling constant $W_4 = W_1 + W_2$ which only depends on the strong Raman and IR transitions (corresponding to one half of the E/A separation) and allows for a safer comparison between theory and experiment. The experimental error of $W_4$ may be conservatively estimated to be around ±2 cm$^{-1}$ as one half of the sum of the Raman calibration error of ±1 cm$^{-1}$ and possible differences between band centers and band maxima in the IR and Raman spectra of up to 3 cm$^{-1}$, which should however largely cancel in the spectral difference.

Depending on the spectral assignments, there are three different ways to analyze the coupling multiplet in the water tetramer. If one neglects couplings beyond the nearest neighbors ($W_2$), the direct coupling constant based on the most active A-symmetric (Raman) and E-symmetric (IR) vibrations is $W_4 \approx W_4 = 34$ cm$^{-1}$, which is to be compared to 12–13 cm$^{-1}$ from the most recent and otherwise fairly accurate analytical potential hypersurface. One could argue that the discrepancy stems from the neglect of next-nearest neighbor coupling. Indeed, inclusion of the weakly Raman-active B-transition at 3438 cm$^{-1}$ reduces the experimental nearest-neighbor coupling
nol tetramers.19 If one does not want to rely on the less certain
but liable to some perturbations in the CH3OD case.19 Upon deutera-
tion, the coupling constants shrink by 30–40%, as in methanol. 19
Dependent on the isotopic composition and analysis method, the methanol coupling is seen to be
10–60% larger than the water coupling, in line with qualitative
and quantitative expectations on the strength of the hydrogen bond and susceptible
to some perturbations in the CH3OD case.19 Upon deuteration,
the coupling constants shrink by 30–40%, as in methanol.19

In view of the involved simplifications, the agreement between experiment and harmonic ab initio calculations is remarkably
good for the (H2O)4 coupling constants. While the B3LYP
calculations43 somewhat overestimate the coupling, available
BSSE-corrected MP2 and higher level electron correlation
methods agree with experiment within the uncertainties of the latter and vibrational CI treatment does not change the
agreement between theory, because such a second-nearest neighbor interaction is essentially absent from the most recent
analytical potential energy hypersurfaces. In contrast, even
simple harmonic ab initio calculations yield largely correct
orders of magnitude, also in line with experiments for metha-
nol tetraturmers.19 If one does not want to rely on the less certain
position of the B-band, one evaluates the sum \( W_1 + W_2 \) defined
above as the B-invariant tetramer coupling \( W_4 \), which corre-
sponds to one half of the IR/Raman band splitting. It is
equivalent to \( W_4 \) only if the diagonal coupling \( W_2 \) is negligible.

Table 2 summarizes the experimental findings for the different
assumptions and isotopologues and compares them to the
methanol findings. Dependent on the isotopic composition
and analysis method, the methanol coupling is seen to be
10–60% larger than the water coupling, in line with qualitative
expectations on the strength of the hydrogen bond and susceptible
to some perturbations in the CH3OD case.19 Upon deuteration,
the coupling constants shrink by 30–40%, as in methanol.19

The present experiments for the first time provide solid evidence
that the coupling among the OH oscillators, and thus resonant energy flow
along the OH stretching modes, is underestimated by at least a factor of two in the most accurate analytical fits. Since the multi-
dimensional fits are truly challenging and the investigated quant-
ity is relatively subtle, this is not an unexpected performance, but
the present experiments for the first time provide solid evidence
and incentive to improve the flexibility of this particular coordinate
coupling. However, in order to exclude a coincidental mismatch,
it is necessary to turn to the more difficult trimer case.

3.3 Water trimer
For the cyclic trimer (ref. 4 and 47 and references cited therein)
one has to distinguish between a C3 symmetric aggregate (simple
Hückel model) where coupling constants between the three
oscillators are equal and an asymmetric coupling case where
the H-bonds differ in strength.19 For the first case the solutions

\[
\begin{bmatrix}
  x & W & W \\
  W & x & W \\
  W & W & x \\
\end{bmatrix} = 0 \tag{3}
\]

are:

\[
x_1 = -2W_4 \quad (A)
\]

\[
x_{2,3} = W \quad (E) \tag{4}
\]

Therefore the coupling constant \( W \) corresponds to the split-
ing between the Raman and IR active levels divided by 3.

In the latter case one obtains two different coupling constants
\( W_1, W_2 \) and the determinant equation

\[
\begin{vmatrix}
  x & W_1 & W_2 \\
  W_1 & x & W_1 \\
  W_2 & W_1 & x \\
\end{vmatrix} = 0 \tag{5}
\]
Its solutions are:

\[
\begin{align*}
x_1 &= -\frac{\sqrt{W_1^2 + W_2^2}}{2} - \frac{W_2}{2} \\
x_2 &= W_2 \\
x_3 &= \frac{\sqrt{W_1^2 + W_2^2}}{2} - \frac{W_2}{2}
\end{align*}
\]

The splitting between the highest and lowest level is therefore \(\sqrt{W_1^2 + W_2^2}\) and the inner level is separated from the center of the multiplet by \(W_2\).

The experimental infrared OH stretching spectrum of the water trimer has been studied in different environments.\(^4,10,33,48,49\) It consists of a main band and a satellite, which is often, but not always located at higher frequency. Upon deuteration, this satellite peak disappears, hinting at a possible tunneling origin. Fig. S8 in the ESI\(^+\) discusses the phenomenological situation in comparison to the tetramer. Comparison between experiment and theory depends on whether one interprets the weak satellite in the IR as one of the near-degenerate OH stretching modes or not. The maximum spread of the bound OH stretching modes is either 57 cm\(^{-1}\) or else \(\approx 42\) cm\(^{-1}\). A more likely value is in between, if one interprets the center of gravity of the two IR bands as the proper reference. On the theoretical side, two fit variants of the three body potential yield a spread of the bound OH stretching modes of 50 vs. 29 cm\(^{-1}\) for the water trimer.\(^5^0\) Matrix isolation IR studies\(^4\) find a large sensitivity of the trimer stretching spectra and their intensities to the environment. The Ne matrix isolation value for the exciton splitting\(^5\) of 57 cm\(^{-1}\) is consistent with our gas phase findings, in particular considering actual matrix-shifts of 15–19 cm\(^{-1}\) in this case.

The non-equivalence of the three hydrogen bonds in the trimer can be included in the analysis and adds further to the uncertainty. In Table 3, we list the experimentally derived trimer coupling constants according to the symmetric \((W=W3)\) and the asymmetric models \((W_1, W_2)\). For the regular hydrogen isotope, we distinguish between the interpretation of the main IR peak as the quasidegenerate E band (case \(3_a\)) and the interpretation of both observed peaks as the components of the E band despite their different intensities and large splitting (less likely case \(3_b\)). Despite the assignment uncertainty, one can see that the W coupling constants show the expected trend with deuteration (decrease) and methyl group substitution (increase), as in the case of the tetramer. When comparing to theory, we always interpret the (now equally intense and more closely spaced) IR doublet in the sense of model \(3_b\) when extracting a symmetric coupling constant \(W\). The performance of harmonically predicted coupling constants is again remarkable, quite independent of the level of computation, as Table 3 shows. The coupling pattern appears to be rather robust, as expected for a transition dipole or hydrogen bond mediated interaction. This is also the case for one of the analytical WHBB (Wang/Huang/Braams/Bowman) fits to the high level \(ab\) \(\text{initio}\) data, namely the variant \(3b6\) for the three body contribution.\(^5^0\) On the other hand, the \(3b5\) fit only recovers about half of the coupling. As this parametrization appears to have been used for the tetramer predictions,\(^2^6,5^4\) this may contribute to the discrepancies discussed in the previous section.

### 3.4 Tentative pentamer analysis

For the cyclic water pentamer, the absence of symmetry\(^5^2\) complicates the coupling analysis and some assumptions are necessary to analyze the experimental coupling pattern. We assume an averaged effective \(C_{5v}\) symmetry, instead of the puckered non-planar ring. Based on simple connectivity considerations, the nearest and next-nearest neighbor excitonic splittings are determined from the following equation:

\[
x W_1 W_2 W_2 W_1 W_1 W_2 W_1 W_1 W_2 x
\]

\[W_1 W_2 W_1 W_1 W_2 x = 0\]

Table 3 | Different theoretical values for the coupling constant \(W3 = W\) for a \(C_{5v}\)-symmetric trimer (the simple Hückel model, one level with A-symmetry (Raman active) and two degenerate levels of E-symmetry (IR-active)) and coupling constants \(W_1\) and \(W_2\) for the more complex case where the hydrogen bonds differ in strength (asymmetric coupling). Experimental trimer coupling constants are less certain than tetramer constants because of the reduced symmetry, the potentially large tunneling splitting, and the intensity difference of the two observed IR-active bands

<table>
<thead>
<tr>
<th>Method</th>
<th>Variant</th>
<th>Electronic structure</th>
<th>(W/cm^{-1})</th>
<th>(W_1/cm^{-1})</th>
<th>(W_2/cm^{-1})</th>
</tr>
</thead>
<tbody>
<tr>
<td>Harmonic(^44)</td>
<td>BSSE-corrected</td>
<td>MP2/6-311+G(2d,2p)</td>
<td>17</td>
<td>19</td>
<td>13</td>
</tr>
<tr>
<td>Vibrational CI(^44)</td>
<td>BSSE-corrected</td>
<td>MP2/6-311+G(2d,2p)</td>
<td>19</td>
<td>21</td>
<td>16</td>
</tr>
<tr>
<td>Vibrational CI(^44)</td>
<td>With BSSE</td>
<td>MP2/6-311+G(2d,2p)</td>
<td>24</td>
<td>26</td>
<td>20</td>
</tr>
<tr>
<td>Harmonic(^42)</td>
<td>PES(1,2,3)</td>
<td>CCSD(T)/aug-cc-pVTZ</td>
<td>11</td>
<td>13</td>
<td>7</td>
</tr>
<tr>
<td>Harmonic(^42)</td>
<td>CP-CCSD(T)/aug-cc-pVTZ</td>
<td>20</td>
<td>22</td>
<td>16</td>
<td></td>
</tr>
<tr>
<td>Harmonic(^42)</td>
<td>CCSD(T)/aug-cc-pVTZ</td>
<td>18</td>
<td>20</td>
<td>14</td>
<td></td>
</tr>
<tr>
<td>Harmonic(^42)</td>
<td>LCCSD(T)/aug-cc-pVTZ</td>
<td>15</td>
<td>16</td>
<td>11</td>
<td></td>
</tr>
<tr>
<td>Harmonic(^5^0)</td>
<td>LCCSD(T)/aug-cc-pVQZ</td>
<td>21</td>
<td>23</td>
<td>17</td>
<td></td>
</tr>
<tr>
<td>Harmonic(^5^0)</td>
<td>3b5 fit</td>
<td>WHBB-surface</td>
<td>8</td>
<td>10</td>
<td>4</td>
</tr>
<tr>
<td>Harmonic(^5^0)</td>
<td>3b6 fit</td>
<td>WHBB-surface</td>
<td>15</td>
<td>17</td>
<td>10</td>
</tr>
<tr>
<td>Anharmonic(^45)</td>
<td>MP2 VPT2 est.</td>
<td>CCSD(T)/aug-cc-pVDZ</td>
<td>18</td>
<td>20</td>
<td>15</td>
</tr>
<tr>
<td>Experiment H(_2)O (3(_a))</td>
<td>IR + Raman</td>
<td>Non-Born–Oppenheimer</td>
<td>14</td>
<td>14</td>
<td>14</td>
</tr>
<tr>
<td>Experiment H(_2)O (3(_a))</td>
<td>IR + Raman</td>
<td>Non-Born–Oppenheimer</td>
<td>17</td>
<td>20</td>
<td>9</td>
</tr>
<tr>
<td>Experiment D(_2)O</td>
<td>IR + Raman</td>
<td>Non-Born–Oppenheimer</td>
<td>11</td>
<td>11</td>
<td>11</td>
</tr>
<tr>
<td>Experiment C(_2)H(_5)OH(^19)</td>
<td>IR + Raman</td>
<td>Non-Born–Oppenheimer</td>
<td>20</td>
<td>21</td>
<td>17</td>
</tr>
<tr>
<td>Experiment C(_2)H(_3)OD(^19)</td>
<td>IR + Raman</td>
<td>Non-Born–Oppenheimer</td>
<td>13</td>
<td>14</td>
<td>11</td>
</tr>
</tbody>
</table>
The solutions to this determinant equation \( x_i \) relative to the center of gravity of the multiplet are (with \( g = 5 - 1 = (\sqrt{5} - 1)/2 \) as the golden ratio):

\[
\begin{align*}
x_1 &= -2W_1 - W_2 \\
x_{2,5} &= -gW_1 + W_2/g \\
x_{4,5} &= W_1/g - gW_2
\end{align*}
\]

The splitting between the dominant IR band \( (E_1') \) and the dominant Raman band \( (A') \) is thus \((2 - g)W_1 + (3 + g)W_2\), whereas the splitting between the two Raman active bands is \((3 + g)W_1 + (2 - g)W_2\). In a puckered ring, the degenerate states will be further split by a small amount. Because the highest frequency transition is not predicted to be strong in either IR or Raman spectra, it is desirable to have a linear combination of coupling constants \( W_5 \) which does not depend on its assignment. This is given by 

\[
W_5 = W_1 + \frac{\sqrt{5} + 1}{\sqrt{5} - 1} W_2
\]

Turning now to the experimental spectra, the dominant IR/Raman water pentamer signals (Fig. 2 and 3) are closer together than the tetramer peaks, despite a reduced ring strain and enhanced cooperativity in the cyclic hydrogen bond pattern. Only the deuterated water spectra offer an assignment of all three bands, if the band at 2500 cm\(^{-1}\) marked >3b is assigned to a dominant ring cluster contribution. In Ne matrices, a blue-shifted\(^53\) trimer bending overtone has been reported\(^5\) at 3190 cm\(^{-1}\). Normally, one would expect much less Raman intensity for such a cluster transition, due to the dominance of monomers in the expansions. Furthermore, a coincidence of tetramer and pentamer signals is unlikely in an unperturbed picture.\(^43\) The progressively smaller energy gap between the OH stretching fundamentals \((4b, 5b)\) and the OH bending overtone \((>3b)\) explains both anomalies by an increasing extent of Fermi resonance between the two states (see Fig. 1), relative to the trimer. The pentamer may have a somewhat higher bending overtone visibility and experience a reduced overtone frequency due to this anharmonic interaction. That is particularly true for the highest frequency concerted bending modes, which are expected to couple most to the Raman active concerted OH stretching modes (Fig. 1).

As a consequence, the latter are shifted to higher frequency, more so in the pentamer than in the tetramer. Indeed, simple model calculations show that the energy coincidence is close in the pentamer. This is illustrated in Table 5, which uses published density functional data\(^43\) to predict the energy coincidence between the experimental stretching fundamental \( nb \) and the anharmonic bend overtones. The latter are estimated from the experimental monomer bend overtone and the calculated complexation shift in the fundamental region. Fermi resonance is more likely if the depolarization ratio is low (like in the

<table>
<thead>
<tr>
<th>Method</th>
<th>Variant</th>
<th>Electronic structure</th>
<th>( W_4/cm^{-1} )</th>
<th>( W_5/cm^{-1} )</th>
<th>( W_5 = W_1 + \frac{\sqrt{5} + 1}{\sqrt{5} - 1} W_2/cm^{-1} )</th>
</tr>
</thead>
<tbody>
<tr>
<td>Harmonic(^52)</td>
<td></td>
<td>MP2/aug-cc-pVDZ</td>
<td>35</td>
<td>9</td>
<td>58</td>
</tr>
<tr>
<td>Harmonic(^43)</td>
<td>Scaled</td>
<td>B3LYP/d-aug-cc-pVDZ</td>
<td>34</td>
<td>10</td>
<td>60</td>
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<tr>
<td>Harmonic(^44)</td>
<td>BSSE-corrected</td>
<td>MP2/6-311+G(2d,2p)</td>
<td>26</td>
<td>7</td>
<td>45</td>
</tr>
<tr>
<td>Vibrational CI(^{44})</td>
<td>BSSE-corrected</td>
<td>MP2/6-311+G(2d,2p)</td>
<td>26</td>
<td>6</td>
<td>42</td>
</tr>
<tr>
<td>Vibrational CI(^{44})</td>
<td>With BSSE</td>
<td>MP2/6-311+G(2d,2p)</td>
<td>43</td>
<td>34</td>
<td>133</td>
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<tr>
<td>Harmonic(^26)</td>
<td>Full-dimensional</td>
<td>WHBB</td>
<td>14</td>
<td>0</td>
<td>13</td>
</tr>
<tr>
<td>Harmonic(^26)</td>
<td>LMon-( \beta ) _dimer</td>
<td>WHBB</td>
<td>14</td>
<td>0</td>
<td>14</td>
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<tr>
<td>Anharmonic(^26)</td>
<td>LMon-( \beta ) _dimer</td>
<td>WHBB</td>
<td>15</td>
<td>0</td>
<td>14</td>
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<tr>
<td>Harmonic(^47)</td>
<td></td>
<td>CCSD[T]/aug-cc-pVDZ</td>
<td>30</td>
<td>9</td>
<td>53</td>
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<tr>
<td>Anharmonic(^45)</td>
<td>MP2 VPT2 est.</td>
<td>CCSD[T]/aug-cc-pVDZ</td>
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<td>21</td>
<td>90</td>
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<tr>
<td>Experiment ( H_2O )</td>
<td>IR + Raman</td>
<td>Non-Born–Oppenheimer</td>
<td>—</td>
<td>—</td>
<td>33</td>
</tr>
<tr>
<td>Harmonic(^52)</td>
<td>( D_2O )</td>
<td>MP2/aug-cc-pVDZ</td>
<td>24</td>
<td>8</td>
<td>44</td>
</tr>
<tr>
<td>Experiment ( D_2O )</td>
<td>IR + Raman</td>
<td>Non-Born–Oppenheimer</td>
<td>14</td>
<td>4</td>
<td>25</td>
</tr>
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</table>
resonance coupling parameter (smaller in the D\textsubscript{2}O case), but the redistribution lifetime\textsuperscript{29} for the tetramer after Raman-active OH overtone is also responsible for the short predicted energy mentally derived statement. The close vicinity of the bending multidimensional nature of the coupling precludes a firm experi-

than >3\textsubscript{o} and positive peaks result from cluster sizes larger than >3\textsubscript{o}. That the >3\textsubscript{o} band cancels, negative peaks emerge for cluster sizes smaller decreasing cluster content from each other after scaling in such a way that the >3\textsubscript{o} band cancels, negative peaks emerge for cluster sizes smaller than >3\textsubscript{o} and positive peaks result from cluster sizes larger than >3\textsubscript{o}.

Table 5 Calculated ring cluster bending wavenumbers $\omega_{bend}^R$ and Raman depolarization ratios $\rho$ are used to estimate the gap $\delta(b-o) = \nu(n\omega_{bend}^o) - \nu(\delta\omega_{bend}^o) = [\nu(n\delta\omega_{bend}^o) + 2(\omega_{bend}^o - \omega_{bend}^{2\omega_{bend}})]$ between the lowest frequency bonded OH stretching fundamental $\omega_{bend}^o$ and the overtones of the bending fundamentals.

<table>
<thead>
<tr>
<th>Cluster size</th>
<th>$\omega_{bend}^R$/cm\textsuperscript{-1}</th>
<th>$\rho$</th>
<th>$\delta(b-o)$/cm\textsuperscript{-1}</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>1622</td>
<td>0.68</td>
<td>504</td>
</tr>
<tr>
<td>3</td>
<td>1658</td>
<td>0.39</td>
<td>266</td>
</tr>
<tr>
<td></td>
<td>1634</td>
<td>0.64</td>
<td>314</td>
</tr>
<tr>
<td></td>
<td>1631</td>
<td>0.75</td>
<td>320</td>
</tr>
<tr>
<td>4</td>
<td>1681</td>
<td>0.25</td>
<td>63</td>
</tr>
<tr>
<td></td>
<td>1651</td>
<td>0.75</td>
<td>123</td>
</tr>
<tr>
<td></td>
<td>1636</td>
<td>0.75</td>
<td>153</td>
</tr>
<tr>
<td>5</td>
<td>1689</td>
<td>0.17</td>
<td>23</td>
</tr>
<tr>
<td></td>
<td>1681</td>
<td>0.05</td>
<td>39</td>
</tr>
<tr>
<td></td>
<td>1660</td>
<td>0.74</td>
<td>81</td>
</tr>
<tr>
<td></td>
<td>1652</td>
<td>0.62</td>
<td>97</td>
</tr>
<tr>
<td></td>
<td>1640</td>
<td>0.73</td>
<td>121</td>
</tr>
</tbody>
</table>

lowest frequency OH stretching fundamental) and the gap $\delta(b-o)$ is small. One can conclude that the Raman active pentamer vibration 5\textsubscript{b} is most likely to undergo strong Fermi resonance with a bending overtone 5\textsubscript{o}. The two highest frequency bending modes indeed have the smallest depolarization ratios and the smallest gaps. This explains the apparently small pentamer OH stretch excitonic coupling in a qualitative way. We estimate 20–50 cm\textsuperscript{-1} for the order of magnitude of the Fermi resonance coupling parameter (smaller in the D\textsubscript{2}O case), but the multidimensional nature of the coupling precludes a firm experimentally derived statement. The close vicinity of the bending overtone is also responsible for the short predicted energy redistribution lifetime\textsuperscript{29} for the tetramer after Raman-active OH stretch excitation. As has been pointed out before,\textsuperscript{35} inclusion of such anharmonic interactions is crucial for a correct modeling and the present experimental data can provide valuable constraints. While there are numerous vibrational predictions for water clusters in the literature, few of them include Raman intensities\textsuperscript{41} or anharmonic effects.\textsuperscript{35} It is hoped that anharmonic calculations including Raman intensities will be carried out in the future to compare more accurately with the present experimental spectra.

3.5 Water hexamer

For the water hexamer, our Raman spectra provide polarized band evidence above 3100 cm\textsuperscript{-1} and 3500 cm\textsuperscript{-1} (>3\textsubscript{b}, see also Fig. 4), but the lack of size-selectivity\textsuperscript{24} does not allow for an unambiguous assignment. A ring structure can be excluded based on the comparison with H\textsubscript{2}O droplet spectra\textsuperscript{16} and a dominant book isomer based on theoretical predictions.\textsuperscript{35} The stable prism and cage structures\textsuperscript{12} are both reasonably consistent with our Raman spectra\textsuperscript{42} with a strong signal above 3100 cm\textsuperscript{-1} (2300 cm\textsuperscript{-1} for D\textsubscript{2}O) and a second pile-up of Raman intensity around the trimer band. Weaker Raman signals predicted between the dominant trimer and tetramer bands are consistent with our spectra, but they overlap with $n = 4$, 5 bands.

4 Conclusions

The first spontaneous Raman scattering spectra of small water clusters offer quantitative insights into the coherent energy hopping of localized OH stretching excitations from one molecule to another, when compared to existing IR spectra. This is related to the non-coincidence effect,\textsuperscript{34} which can be spectrally resolved in the case of cold clusters.\textsuperscript{23} Table 6 summarizes the robust OH and OD oscillator coupling constants for the water trimer (W3), the tetramer (W4) and the pentamer (W5). The deuteration and methylation\textsuperscript{19} effects are systematic and expected. The ring size dependence is unusually weak beyond the tetramer, which we explain by the growing importance of stretch-bend Fermi resonance for the symmetric modes. Effects of intra-monomer coupling should be smaller in the case of cyclic water clusters, because the strong hydrogen bonds decouple the free and the bound OH stretching modes quite effectively.

The excitonic coupling constants obtained in this experimental study for cyclic water clusters provide valuable constraints on harmonic,\textsuperscript{20} anharmonic\textsuperscript{26,44} and empirical\textsuperscript{46} models for the

<table>
<thead>
<tr>
<th>M</th>
<th>W3/cm\textsuperscript{-1}</th>
<th>W4/cm\textsuperscript{-1}</th>
<th>W5/cm\textsuperscript{-1}</th>
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</thead>
<tbody>
<tr>
<td>H\textsubscript{2}O</td>
<td>14–17</td>
<td>34</td>
<td>33</td>
</tr>
<tr>
<td>D\textsubscript{2}O</td>
<td>11</td>
<td>22</td>
<td>25</td>
</tr>
<tr>
<td>CH\textsubscript{3}OH\textsuperscript{19}</td>
<td>20</td>
<td>40</td>
<td></td>
</tr>
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</table>
vibrational dynamics in hydrogen-bonded systems and water in particular. In a somewhat model-dependent way, they can be interpreted within a time-dependent framework. Already for the trimer, the splittings are consistent with a drop of any local excitation to 1/e within about 0.3 ps.\textsuperscript{19,27} For tetramers and pentamers, the coherent transfer to neighboring modes will be even faster. In an extended network such as liquid water, this will follow, in particular for cluster sizes beyond the trimer, and ultimately dissipate the energy into low frequency modes.\textsuperscript{29} However, the primary step seems to be explainable in a harmonic picture\textsuperscript{57} of mode couplings and this is consistent with the present observation that harmonically determined coupling constants agree quite well with experiment up to the tetramer. Water potentials used for the modeling of fs dynamics in water\textsuperscript{8,26,50} should thus be tested for their ability to predict the IR/Raman splitting observed in water clusters in this work. This hydrogen bond-induced excitonic splitting is a subtle, yet crucial ingredient for multidimensional cluster potential energy surfaces, in particular for the smallest clusters of the most important molecule on earth.

**Acknowledgements**

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**References**